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6a,7-Dihydro-2,6-dimethyl-7-phenyl-6H-chromeno[4,3-d]-1,2,4-triazolo[1,5-a]-pyrimidin-6a-ol†

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Abstract

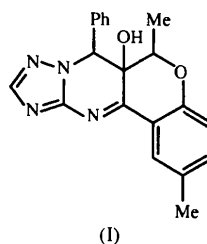
In the title compound, C₂₀H₁₈N₄O₂, the dihydropyrimidine and dihydropyran rings adopt a distorted sofa conformation. The phenyl substituent at C7 and the hydroxy group at C6a adopt axial positions, whereas the methyl group at C6 is equatorial.

Comment

Derivatives of 1,2,4-triazolo[1,5-a]pyrimidine possess high biological activity, for example, as cardiovascular agents (Tsuda *et al.*, 1986). The chemical and physiological characteristics of these compounds are directly related to their conformations. However, relevant data on

† Alternative name: 6a,7-dihydro-2,6-dimethyl-7-phenyl-6H-[1,2,4]-triazolo[2',3':1,2]pyrimidino[5,4-c]chromen-6a-ol.

the structures of dihydro-1,2,4-triazolo[1,5-a]pyrimidines condensed with pyran rings, (I), or other heterocycles have not previously been available.



Torsion angles N7a—C10a—N11—C11a and C1—C1a—C11a—N11 are 15.4 (4) and 9.4 (4)°, respectively. This indicates some loss of planarity of the conjugated system of the molecule, although the C1a—C11a, N11—C11a and C10a—N11 bond lengths [1.453 (4), 1.294 (3) and 1.381 (3) Å, respectively] are evidence of the conjugation between the benzene ring, azomethine group and 1,2,4-triazole ring (Fig. 1).

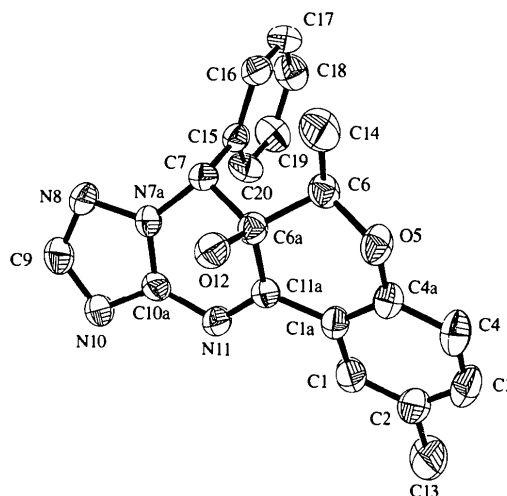


Fig. 1. A view of (I) with displacement ellipsoids drawn at the 50% probability level. The H atoms are omitted for clarity.

The conformations of both the pyrimidine and pyran rings can be described as distorted sofas. The puckering coordinates (Zefirov, Palyulin & Dashevskaya, 1990) are $S = 0.63$, $\theta = 50.9$ and $\varphi = 18.3$ for the pyrimidine, and $S = 0.72$, $\theta = 44.7$ and $\varphi = 10.2$ for the pyran moiety, where S is the puckering degree, and θ and φ are polar angles which describe the ring conformation type (values of these angles for ideal conformations are: sofa $\theta = 45$, $\varphi = 0$; half chair $\theta = 45$, $\varphi = 30$). Comparison of the title compound with *cis*-3-bromo-2,3-dihydro-2-phenyl-4*H*-1-benzopyran-4-one and (2*R*,3*R*)-3,5,7-trihydroxy-2-(2*R**,3*R**)-3-(4-hydroxy-3-methoxyphenyl)-2-hydroxymethyl-(1,4-benzodioxan-6-yl)-4-chromanone

shows that the dihydropyranone ring adopts a distorted half-chair conformation and annelation of the ring concerned leads to the change in its conformation (Cantrell & Hockstein, 1982; Lotter & Wagner, 1983). An alkyl substituent in the second position has an equatorial orientation (Geddes, Sheldrick & Akkrig, 1980; Chambers & Marfat, 1994). Comparison of (I) with 5,7-diphenyl-6-oxy-6,7-dihydropyrazolo[1,5-*a*]pyrimidine, the puckering coordinates for which are $S = 0.61$, $\theta = 46.2$ and $\varphi = 28.9$, indicates that annelation of the dihydropyrimidine ring results in a decrease in its twisting (Desenko *et al.*, 1993).

The crystal is composed of a racemic mixture of the (6*R*,6*aS*,7*R*) and (6*S*,6*aR*,7*S*) isomers. The phenyl and hydroxyl substituents on C7 and C6a are both in axial positions [the C10a—N7a—C7—C15 and O12—C6a—C11a—C1a torsion angles are 95.7 (3) and -104.2 (3)°, respectively]. The methyl group is equatorial; the C4a—O5—C6—C14 and C11a—C6a—C6—C14 torsion angles are -178.7 (2) and -168.1 (2)°, respectively.

In the crystal, the molecules of (I) form chains due to intermolecular hydrogen bonds O12—H12O...N10', characterized by H...N 1.92 (4) Å and O—H...N 174 (4)° [symmetry code: (i) $\frac{1}{2} - x, \frac{1}{2} + y, \frac{3}{2} - z$].

Experimental

Compound (I) was obtained from 2,9-dimethyl-3-phenyl-3,7-dihydro-2*H*-1,2,4-triazolo[1',5'-*a'*]pyrimido[4,5-*d*]benzo[*b*]pyran by dissolving the latter in DMSO and allowing slow aerial oxidation to occur over a period of two weeks. Spectral characteristics of (I) are as follows: UV spectrum (isopropanole, Specord M40 spectrometer): λ_{\max} ($\epsilon/1000$): 313 (20.7), 376 (20.8); IR spectrum (KBr tablets, Specord 75 IR spectrometer): $\nu(\text{C}=\text{N})$ 1618 cm^{-1} ; PMR spectrum (DMSO-*d*₆, Varian VXR-300 spectrometer): δ (p.p.m.): 6.8–7.9 (8H, *m*), 7.94 (1H, *s*, 6-H), 7.18 (1H, *s*, OH), 5.59 (1H, *s*, 3-H), 3.68 (1H, *q*, $J = 6.4$ Hz, 2-H), 2.32 (3H, *s*, 13-CH₃), 1.41 (3H, *d*, $J = 6.4$ Hz, 14-CH₃). Single crystals were grown by diffusion of benzene into a solution of (I) in DMSO.

Crystal data

C₂₀H₁₈N₄O₂
 $M_r = 346.38$
 Monoclinic
 $P2_1/n$
 $a = 10.770$ (2) Å
 $b = 10.045$ (2) Å
 $c = 16.499$ (4) Å
 $\beta = 105.07$ (2)°
 $V = 1723.6$ (6) Å³
 $Z = 4$
 $D_x = 1.335$ Mg m⁻³
 D_m not measured

Data collection

Siemens P3/PC diffractometer

Mo $K\alpha$ radiation
 $\lambda = 0.71073$ Å
 Cell parameters from 24 reflections
 $\theta = 11$ –12°
 $\mu = 0.089$ mm⁻¹
 $T = 293$ (2) K
 Plate
 0.40 × 0.20 × 0.10 mm
 Yellow

$R_{\text{int}} = 0.023$
 $\theta_{\text{max}} = 25.04^\circ$

θ - 2θ scans

Absorption correction: none
 2618 measured reflections
 2467 independent reflections
 1685 reflections with
 $I > 2\sigma(I)$

$h = 0 \rightarrow 12$

$k = -11 \rightarrow 0$

$l = -19 \rightarrow 18$

2 standard reflections
 every 98 reflections
 intensity decay: 5%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.043$
 $wR(F^2) = 0.201$
 $S = 1.152$
 2365 reflections
 241 parameters
 H atoms riding
 $w = 1/[\sigma^2(F_o^2) + (0.0575P)^2 + 0.7176P]$
 where $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.037$

$\Delta\rho_{\text{max}} = 0.17$ e Å⁻³

$\Delta\rho_{\text{min}} = -0.19$ e Å⁻³

Extinction correction: none

Scattering factors from
*International Tables for
 Crystallography* (Vol. C)

Refinement on F^2 was carried out on all reflections except for 66 with very negative F^2 or having $|F_o - F_c| > 4\sigma(F_o)$ and therefore flagged by the user for potential systematic errors. Standard uncertainties on C—C bond distances are 0.004 Å.

Data collection: P3/PC Diffractometer Program (Siemens, 1989). Cell refinement: P3/PC Diffractometer Program. Data reduction: SHELXTL (Sheldrick, 1994). Program(s) used to solve structure: SHELXTL. Program(s) used to refine structure: SHELXTL. Molecular graphics: SHELXTL. Software used to prepare material for publication: SHELXTL.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: BM1134). Services for accessing these data are described at the back of the journal.

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